

Introduction to Reactor System

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Table of Contents

1 QUIZ	3
1.1 <i>QUESTIONS</i>	3
1.2 <i>ANSWERS</i>	3

1 Quiz

1.1 Questions

1. Determine the number of nucleons in $^{238}_{92}\text{U}$.
2. What is the equivalent of 1 J in MeV?
3. Write Einstein's mass-energy relationship.
4. From Einstein's mass-energy relationship, prove that binding energy corresponding to mass defect of 1 amu is approximately 931 MeV.
5. Determine the binding energy of the nucleus in $^{233}_{90}\text{Th}$. The mass of Th-233 is 233.041581 amu.
6. Determine the average binding energy of the nucleus in ^4_2He . The mass of ^4He is 4.002603 amu.
7. Calculate the velocity of a neutron whose kinetic energy is 2 MeV.

1.2 Answers

1. Number of nucleons = number of particles heavier than electron = number of protons + number of neutrons

$$\text{Number of protons} = 92$$

$$\text{Number of neutrons} = 238 - 92 = 146$$

$$\text{Number of nucleons} = 146 + 92 = \underline{\underline{238}}$$

2. $1 \text{ MeV} = 1.6 \times 10^{-13} \text{ J}$

$$\text{Therefore, } 1 \text{ J} = 6.25 \times 10^{12} \text{ MeV}$$

3. $E = mc^2$ (E is the energy, m is the mass and c is the velocity of light)

4. $E = m_d c^2$

$$c = 2.998 \times 10^8 \text{ m/s}; m_d = 1 \text{ amu} = 1.66 \times 10^{-24} \text{ g} = 1.66 \times 10^{-27} \text{ kg}$$

$$E = 1.66 \times 10^{-27} \times (2.998 \times 10^8)^2 = 932.5 \text{ MeV}$$

The difference between the calculated and target values is very small. This could be due to errors in rounding off.

5. Recalling Eq. (2), $m_d = Z(1.007825) + N(1.008665) - M$

$$Z = 90; A = 233; N = A - Z = 143, M = 233.041581 \text{ amu}$$

Substituting above in Eq. (2) yields a mass defect (m_d) of 1.90176 amu.

Recalling that 1 amu corresponds to 931 MeV, the binding energy of U-235 is 1770.5 MeV

6. Recalling Eq. (2), $m_d = Z(1.007825) + N(1.008665) - M$

$$Z = 2; A = 4; N = A - Z = 2, M = 4.002603 \text{ amu}$$

Substituting above in Eq. (2) yields a mass defect (m_d) of 0.030377 amu.

Recalling that 1 amu corresponds to 931 MeV, the binding energy of U-235 is 28.281 MeV

$$\text{Average binding energy} = \text{Binding energy}/A = 28.281/4 = 7.07025 \text{ MeV}$$

7. From Eq. (6), we have $0.5m_n u_n^2 = E_n$

$$m_n = 1.008665 \text{ amu} = 1.6744 \times 10^{-27} \text{ kg}$$

$$E_n = 2 \text{ MeV} = 3.2 \times 10^{-13} \text{ J}$$

Substituting for m_n and E_n in Eq. (6), we get the velocity of neutron as 19550 km/s